Basic principles of multidimensional NMR in solution

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AG Solution NMR

The program

General aspects Basic principles Parameters in NMR spectroscopy Multidimensional NMR-spectroscopy Protein structures NMR-spectra of proteins Sequence specific assignment Protein structure determination Ligand-screening



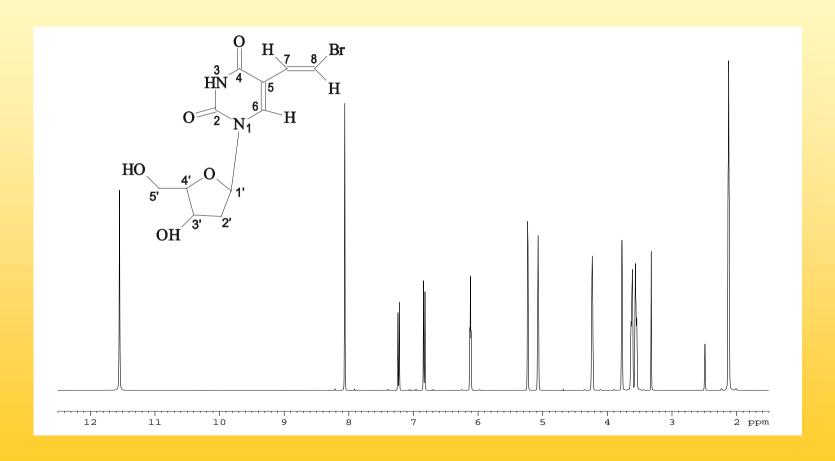


Nuclear Magnetic Resonance

NMR-spectroscopy observes the resonance interaction of atomic nuclei with electromagnetic waves. The effect is only detectable in a strong magnetic field. Every atomic nucleus is observed separately and in addition interactions between nuclei can be visualized. NMR therefore corresponds well to the chemists view of a molecule as atoms connected by bonds.



Analytical method accompanying synthetic work





Structure elucidation of natural compounds

NMR is very powerful in the determination of the constitution of natural products



Determination of the three-dimensional structure of proteins

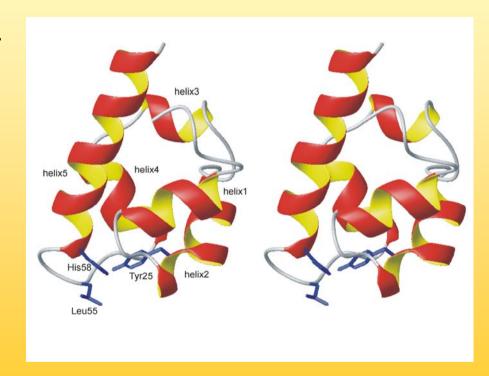
NMR can help to determine

the 3D structure of

proteins at atomic

resolution, in solution as

well as in the solid state





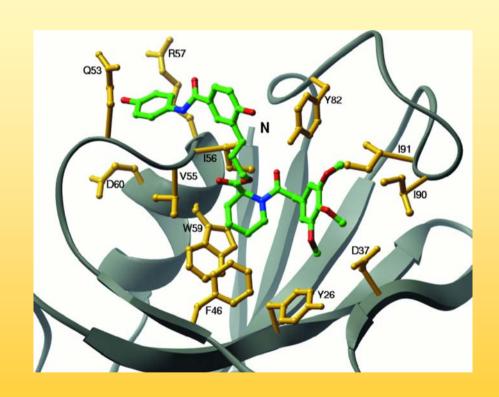
Determination of molecular interactions

NMR can be used to

detect the

interaction between

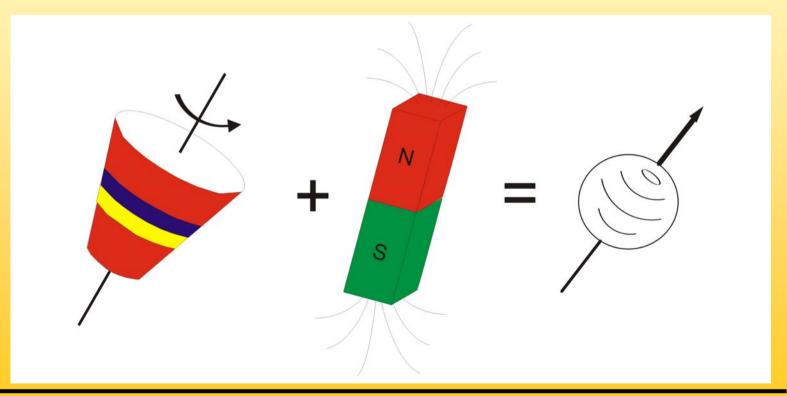
proteins and ligands





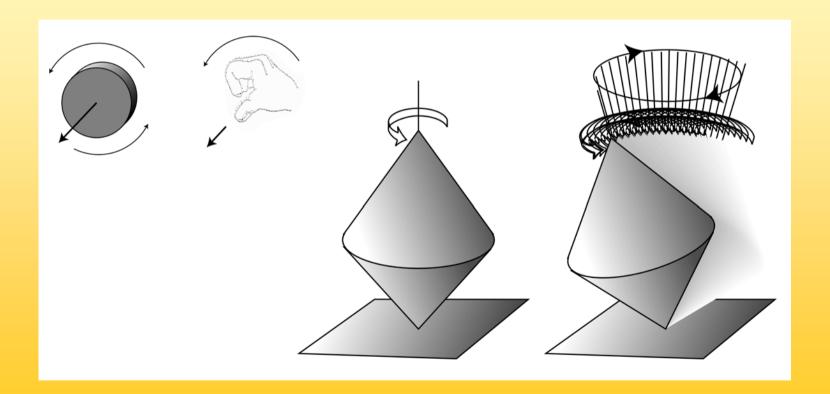


Basis of the effect of nuclear magnetic resonance is the nuclear spin, that can be imagined as a mixture of gyroscope and magnetic needle



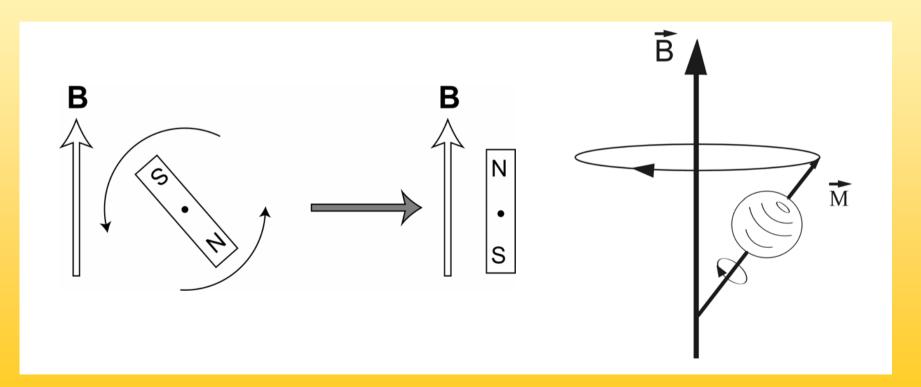


A gyroscope has an angular momentum whose axis is stable in three-dimensional space





An alignment of the "magnetic needle" with an external magnetic field is prevented by the properties of a gyroscope, a precession begins





The resonance frequency of the spins is determined by the magnetic field, as is the sensitivity and the resolution of the spectra

B ₀ [Tesla]	ν ₀ [MHz]		
1.4	60		
5.9	250		
9.4	400		
14.1	600		
21.2	900		



Magnetic properties of relevant nuclei

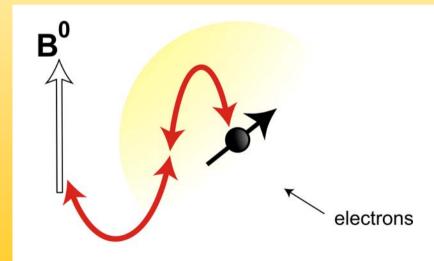
		N		\!\\D_=
Isotop	Spin	Natürliche Häufigkeit	gyromagnetisches	NMR-Frequenz
			Verhältnis g	bei 2.35 T
1H	1/2	99.98	26.7522	100.000
2H	1	0.015	4.1066	15.351
3H	1/2	0	28.5350	106.663
7Li	3/2	92.58	10.3976	38.863
11B	3/2	80.42	8.5847	32.084
12C	0	98.89		
13C	1/2	1.11	6.7283	25.144
14N	1	99.63	1.9338	7.224
15N	1/2	0.37	-2.7126	10.133
170	5/2	0.037	-3.6280	13.557
19F	1/2	100.0	25.1815	94.077
23Na	3/2	100.0	7.0704	26.451
25Mg	5/2	10.13	-1.6389	6.1195
31P	1/2	100.0	10.8394	40.481
35Cl	3/2	75.53	2.6242	9.798
39K	3/2	93.1	1.2499	4.667
43Ca	7/2	0.145	-1.8028	6.728
51V	7/2	99.76	0.052	26.289
57Fe	1/2	2.19	0.8687	3.231
75As	3/2	100.0	4.5961	17.126
77Se	1/2	7.58	5.1214	19.067
113Cd	1/2	12.26	-5.9609	22.182





Chemical shift

Electrons around the nucleus shield it from the external magnetic field, the more electrons the weaker the field



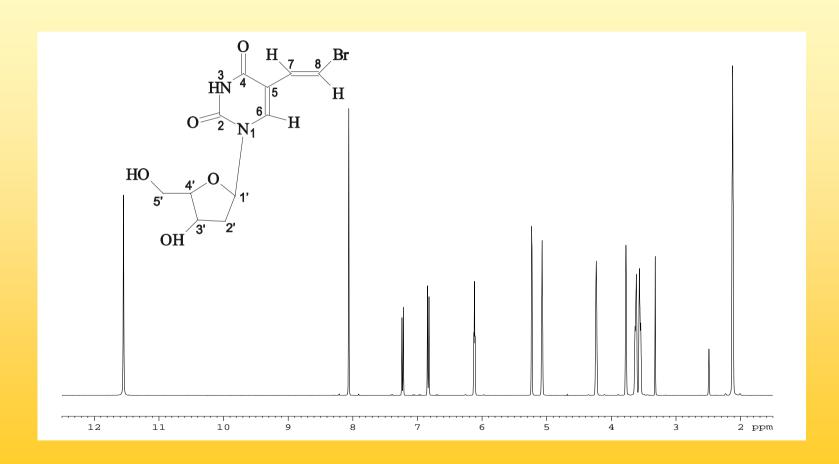
$$B_{eff} = (1 - \sigma) B_0$$

$$\omega = \gamma (1 - \sigma) B_0$$

$$\delta = (\omega - \omega_{ref}) / \omega_0 \times 10^6$$

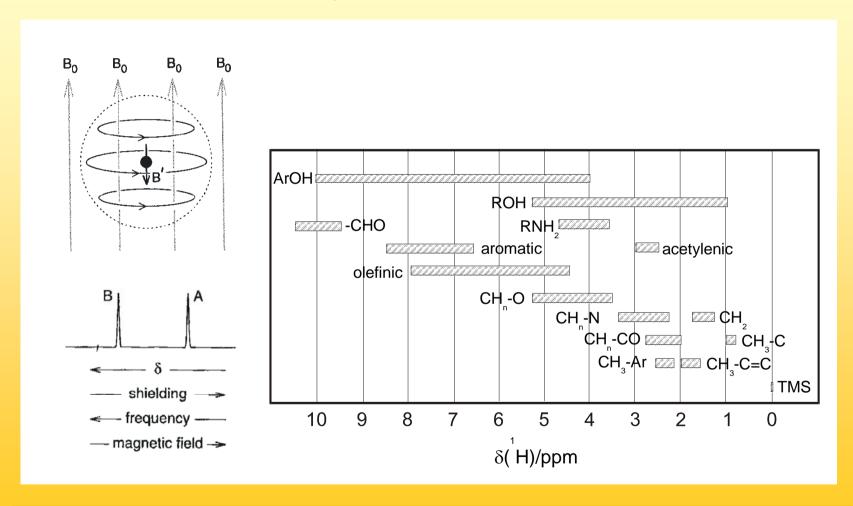
$$= (\sigma_{ref} - \sigma) \times 10^6$$

Each atom in the molecule gives rise to a resonance line

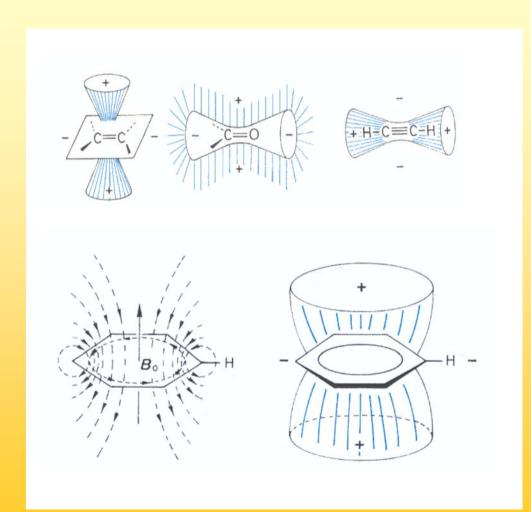




The chemical shift depends on the chemical environment





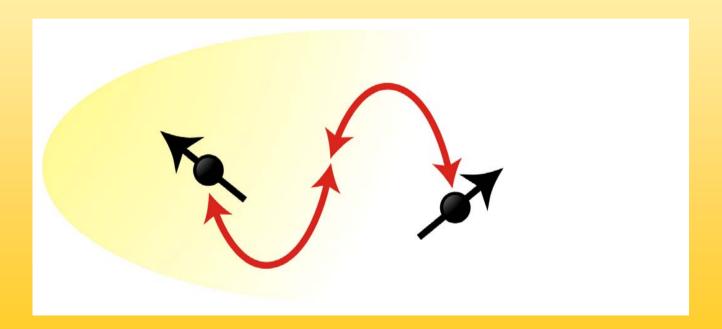


An important factor influencing the chemical shift are anisotropy effects, that are created by small additional fields



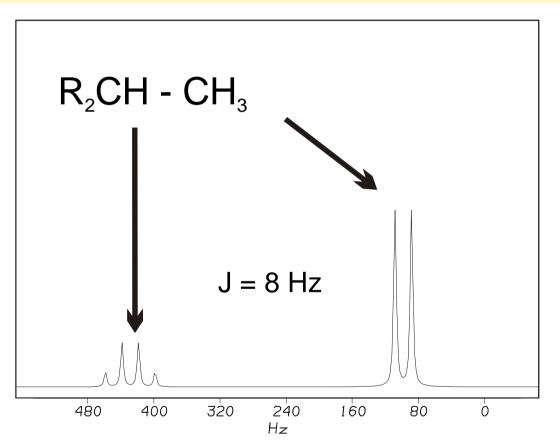
Scalar or J-coupling

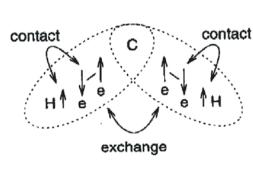
Electrons in the bonds between the nuclei mediate an interaction, the scalar coupling





Scalar coupling splits the signals according to the number of neighboring nuclei

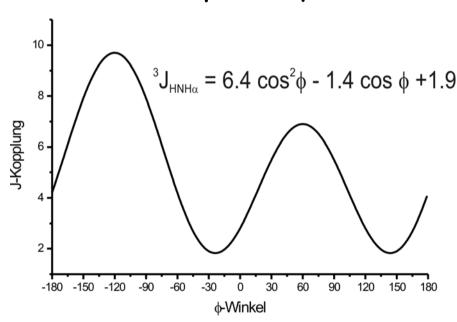


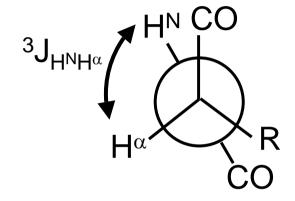


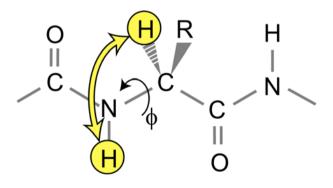


Scalar coupling contains structural information

Karplus-equation



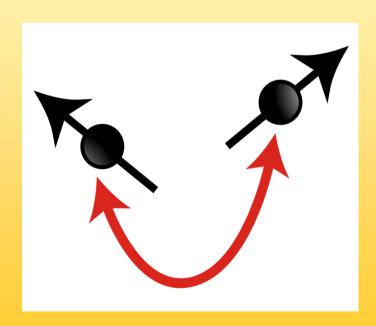






Dipolar coupling

The nuclei interact directly through space via a dipol-dipol interaction



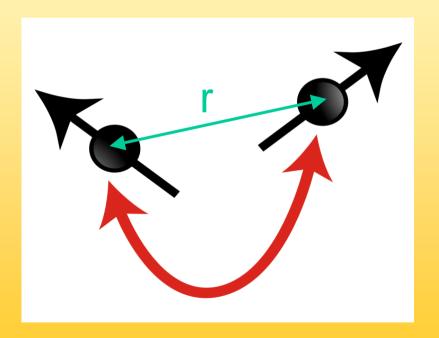
In solution NMR this interaction is averaged to zero due to the fast isotropic movement of the molecules but it is still a source of relaxation



One aspect of relaxation is the NOE-Effect, that depends on the distance between two nuclei

 $I_{NOE} \sim 1/r^6$

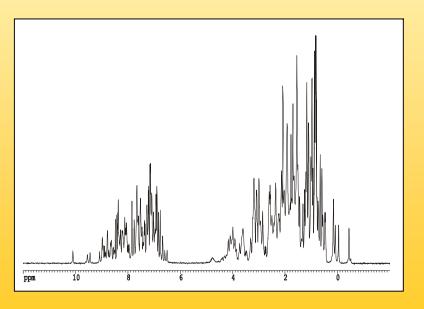
Since the intensity drops quickly with increasing distance the effect can only be observed up to 500 pm

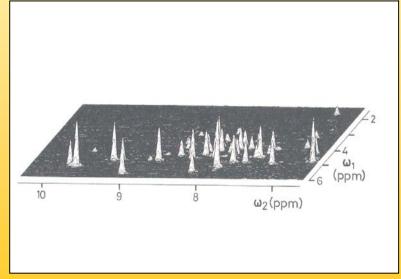






1D-NMR: 2 axis intensity vs. frequency 2D-NMR: 3 axis intensity vs. frequency (1) vs. frequency (2)







The two major advantages of multidimensional NMR are:
Improved resolution: Signals are spread over a surface
(2D) or in a three-dimensional space (3D, 4D)

Magnetization transfer: Signals result from the interaction between nuclei. That can be interactions through bond (via J-coupling) or through space (via NOE).

Taken together this eases the interpretation and the assignment of the spectra considerably



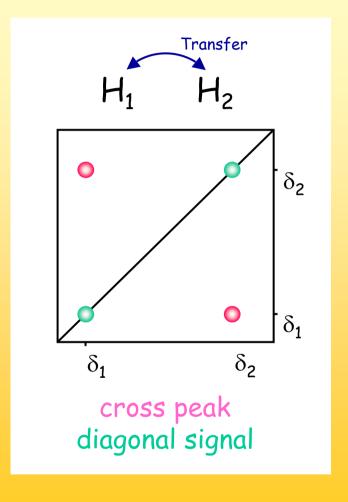
homonuclear spectra

Transfer of magnetization takes place between like nuclei. Both axis exhibit the chemical shift of the same type of nucleus. If a transfer has taken place, the signal has different frequencies in the two dimensions:

cross peak

If no transfer has taken place, the shifts are the same in both dimensions:

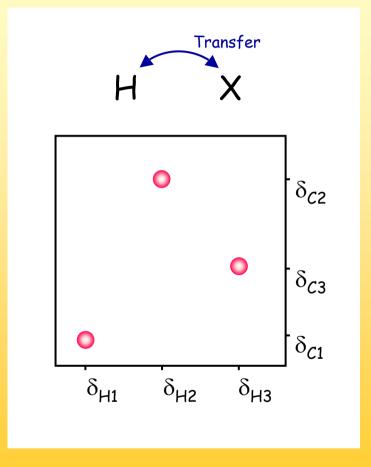
diagonal signal



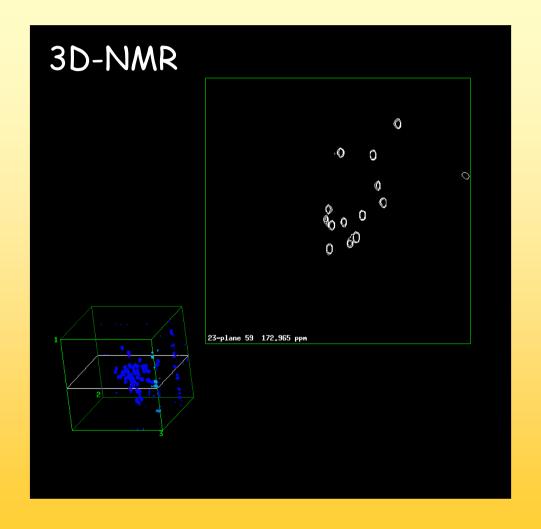


heteronuclear spectra

Transfer of magnetization takes place between nuclei of different types. The two axis show the chemical shift of the respective type of nucleus. If a transfer has taken place, a signal appears at the intersection of the two frequencies, without a transfer there is no signal.



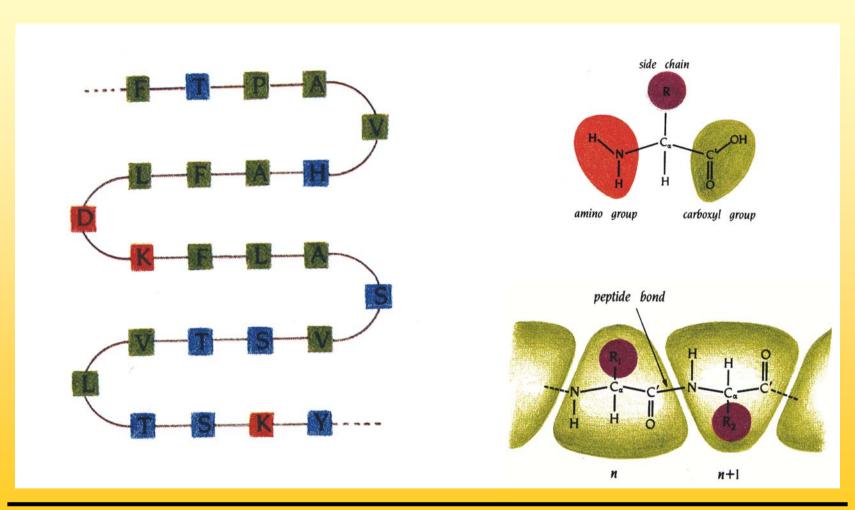






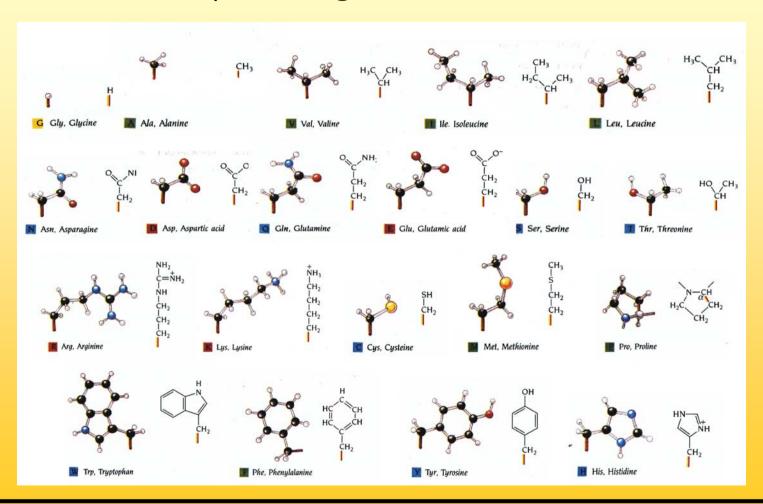


Primary structure



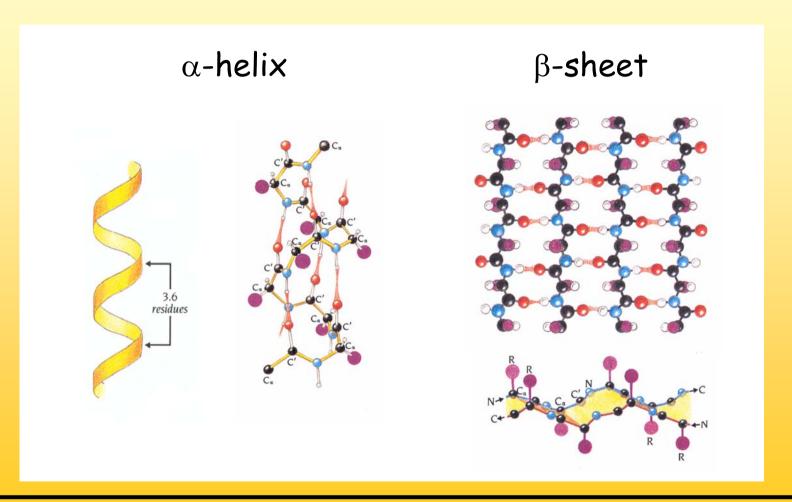


20 proteinogenic amino acids



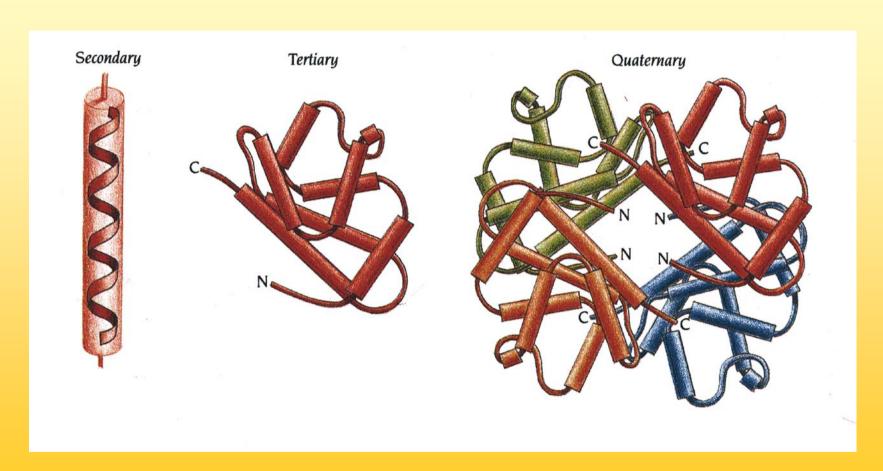


secondary structure





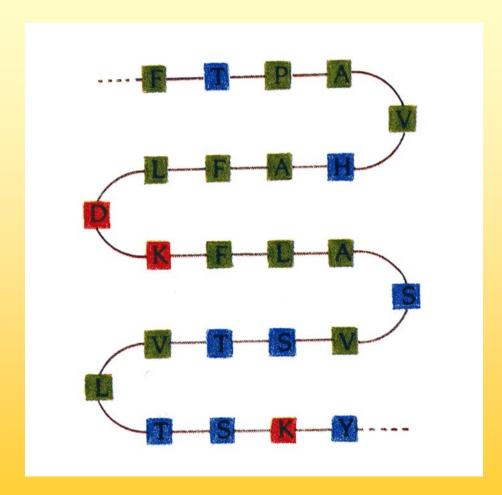
Levels of structural organization





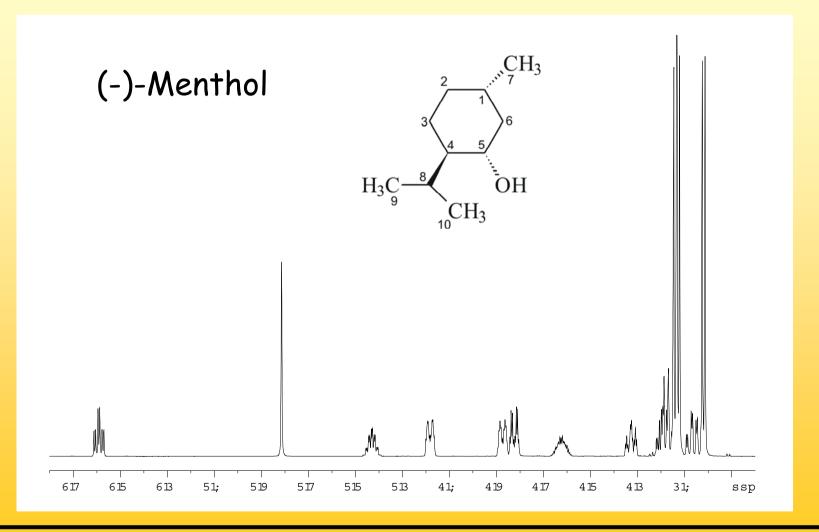
NMR-spectroscopy of proteins



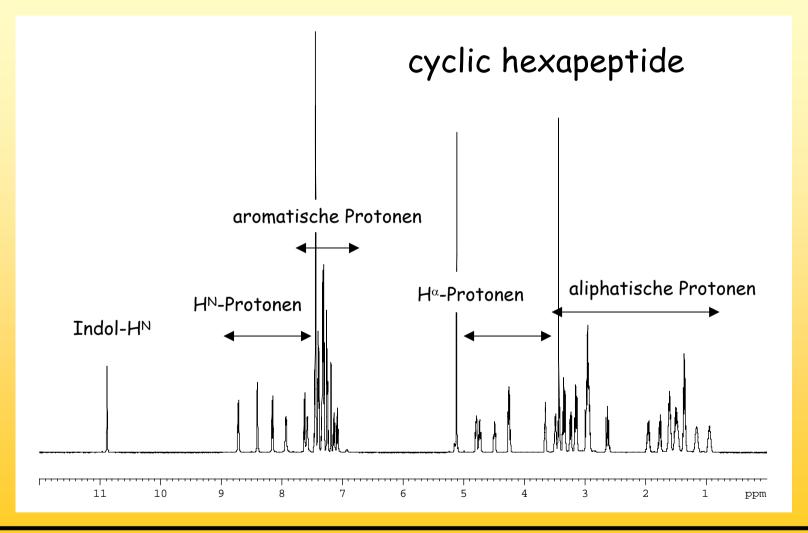


The major problem of protein NMR results from the fact that proteins are polymers, i.e. the repetition of almost identical subunits

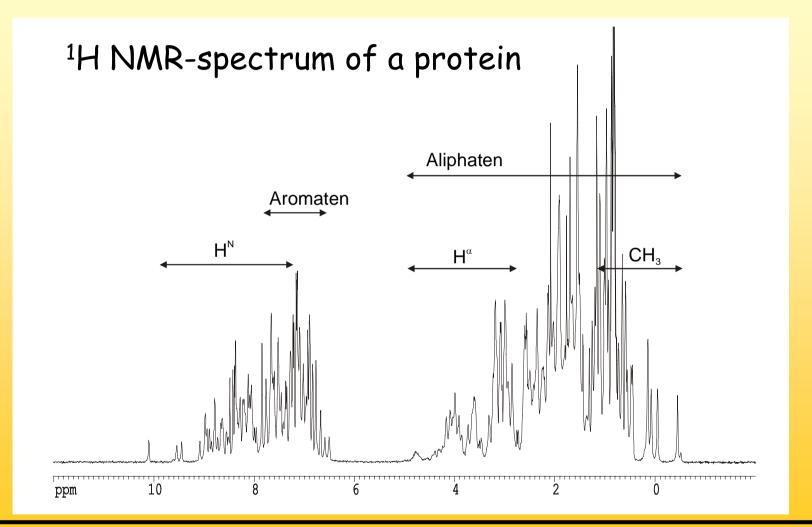






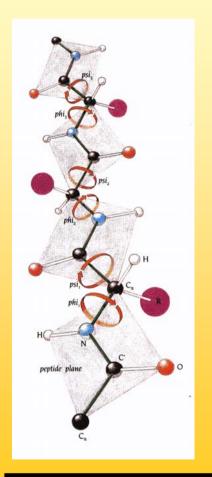


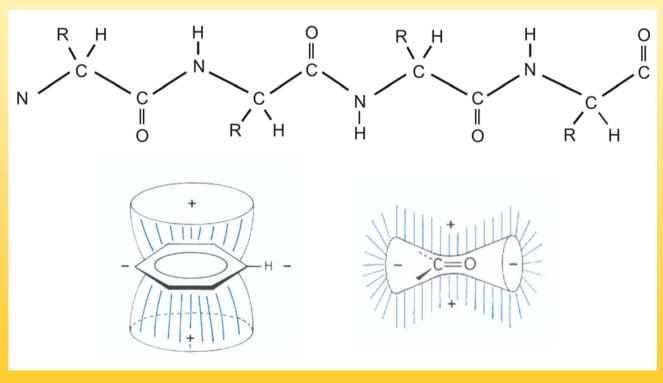






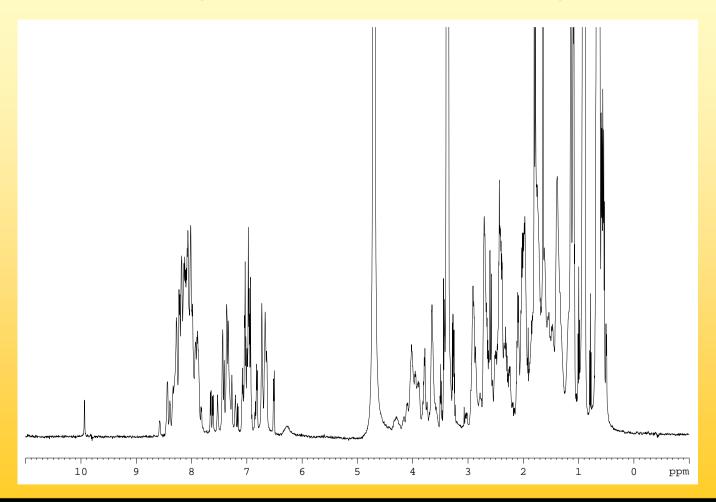
Differences in chemical shifts can be produced by structure and the accompanying anisotropy effect







¹H NMR-spectrum of an unfolded protein







The solution of the assignment problem is the sequence-specific assignment

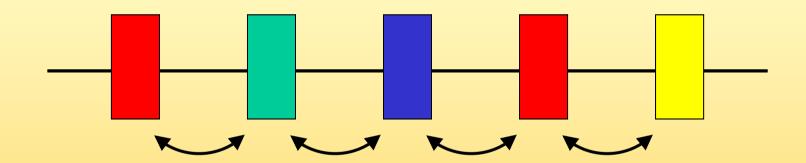
Two strategies exist:

In case of small proteins or peptides where usually only unlabeled material is available the strategy is based on homonuclear spectra (COSY, TOCSY, NOESY)

In case of larger proteins labeling with ¹³C and ¹⁵N is necessary and heteronuclear triple resonance experiments (CBCA(CO)NNH,CBCANNH) are recorded



Sequence-specific assignment

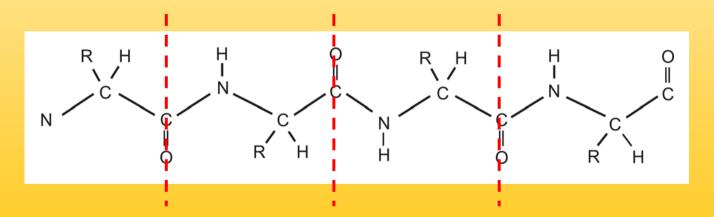


- 1. Which amino acid type is present (which color)
- 2. Which amino acid is next to which (neighborhood)
- 3. Comparison of subsequences with that of the protein



Assignment using homonuclear spectra:

Each amino acid represents a separate set of signals, a spin system, since amino acids are separated by the carbonyl carbon that does not have a proton attached. Homonuclear spectra that utilize scalar couplings (COSY, TOCSY) are used to establish the amino acid type





The neighborhood of the amino acids are then detected by through space interactions, i.e. in NOESY spectra.

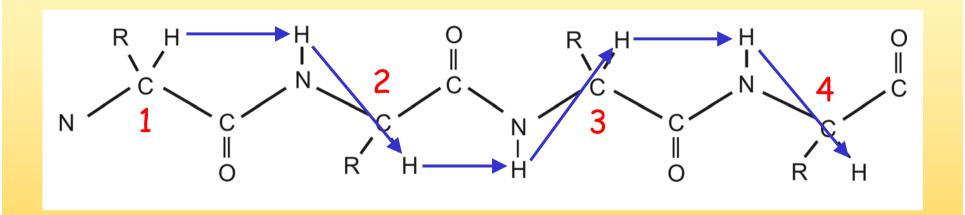
Inter- and intra-residue signals are separated by comparison between the scalar-coupling spectra that can only show intra-residual peaks and the NOESY



The distance from the H^N to the H^α of the same amino acids, $d_{N\alpha}(i,i)$, is always short enough to yield an NOE. The same is true for the distance from the H^N to the H^α of the amino acid (i-1), $d_{\alpha N}$



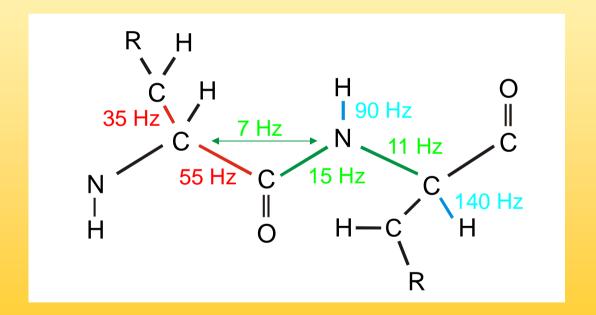
A neighborhood of amino acids is thus established



$$1 \xrightarrow{d_{\alpha N}} 2 \xrightarrow{d_{N\alpha}(i,i)} 2 \xrightarrow{d_{\alpha N}} 3 \xrightarrow{d_{N\alpha}(i,i)} 3 \xrightarrow{d_{\alpha N}} 4 \xrightarrow{d_{N\alpha}(i,i)} 4$$



Triple resonance experiments use the couplings between ¹H, ¹³C und ¹⁵N

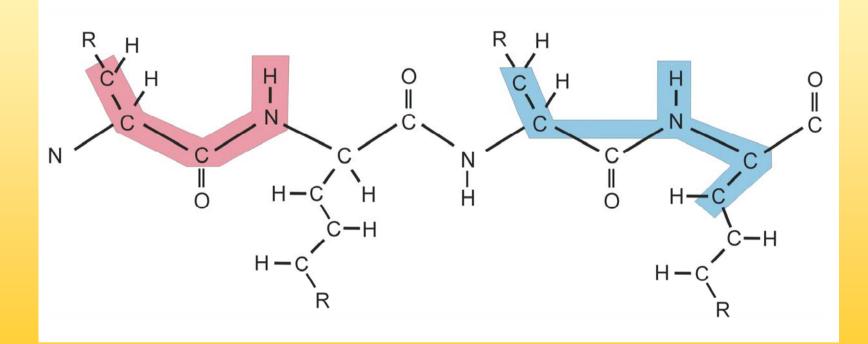




Mainchain assignment using tripel resonance experiments

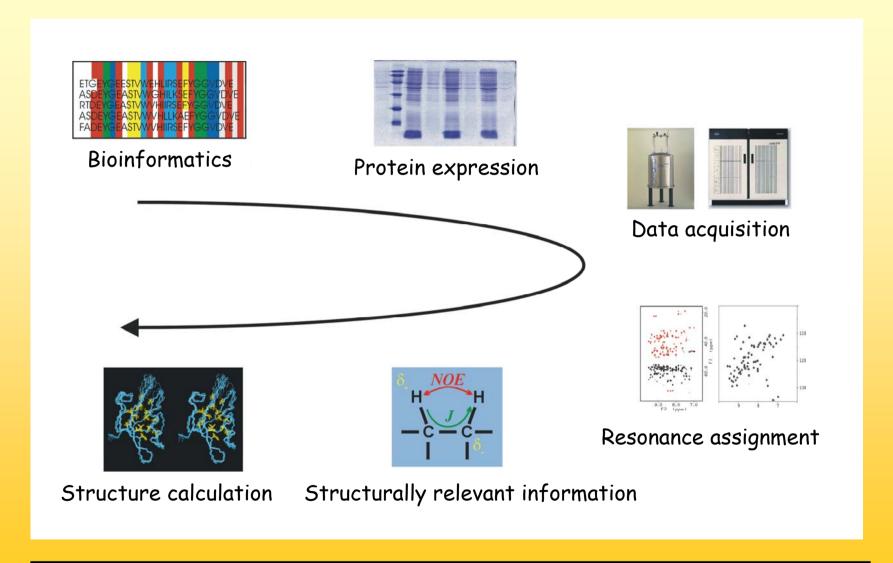
CBCA(CO)NNH

CBCANNH



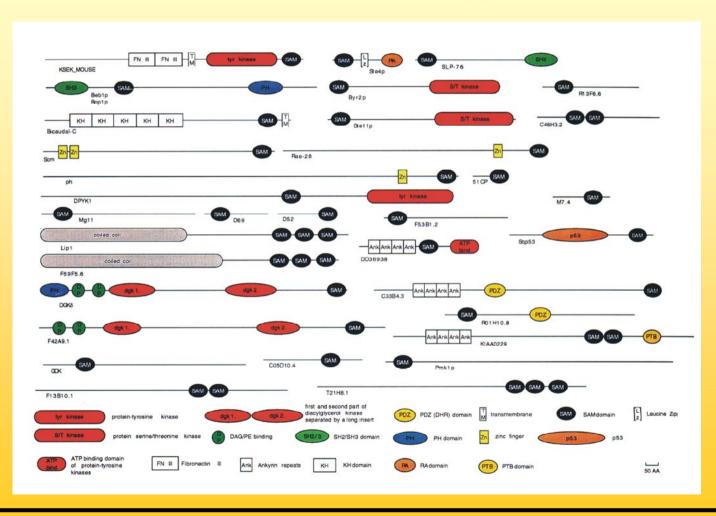
A structure determination using NMR-spectroscopy





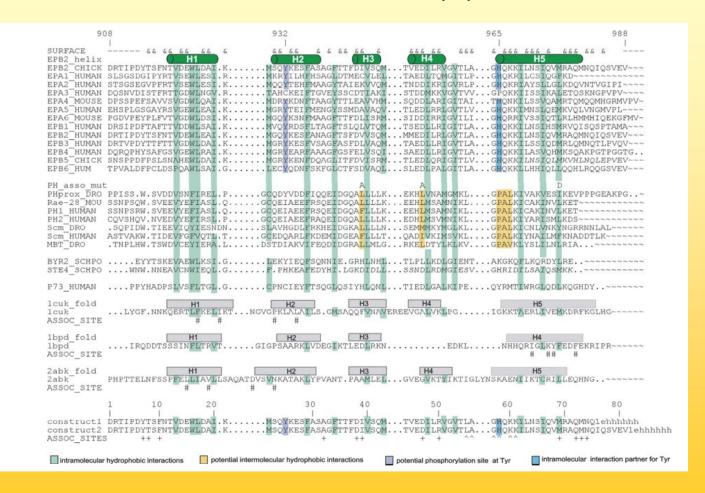


Bioinformatics (1)



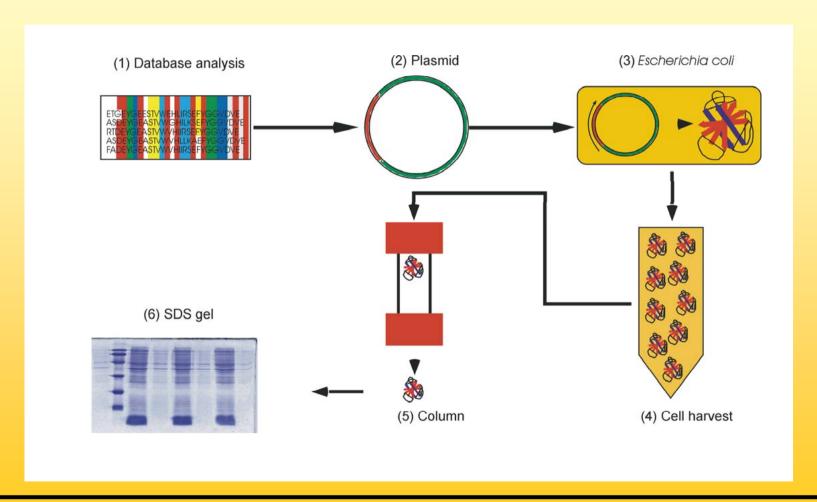


Bioinformatics (2)

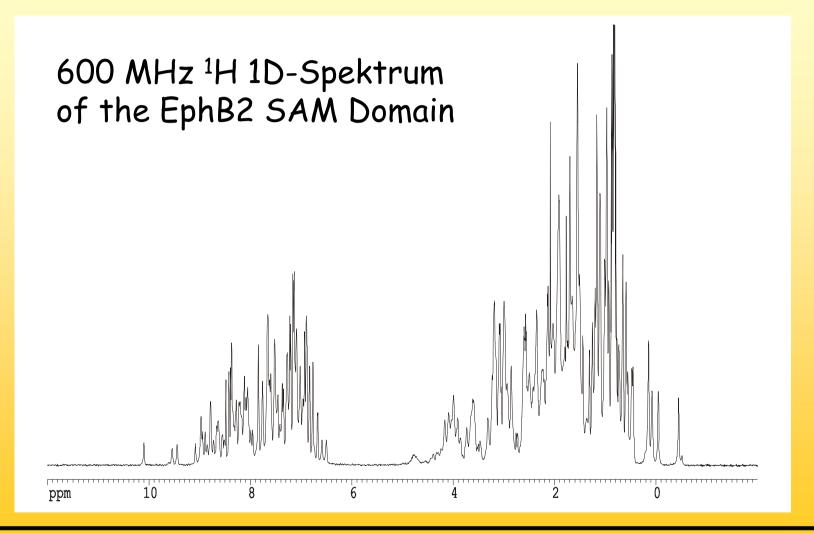




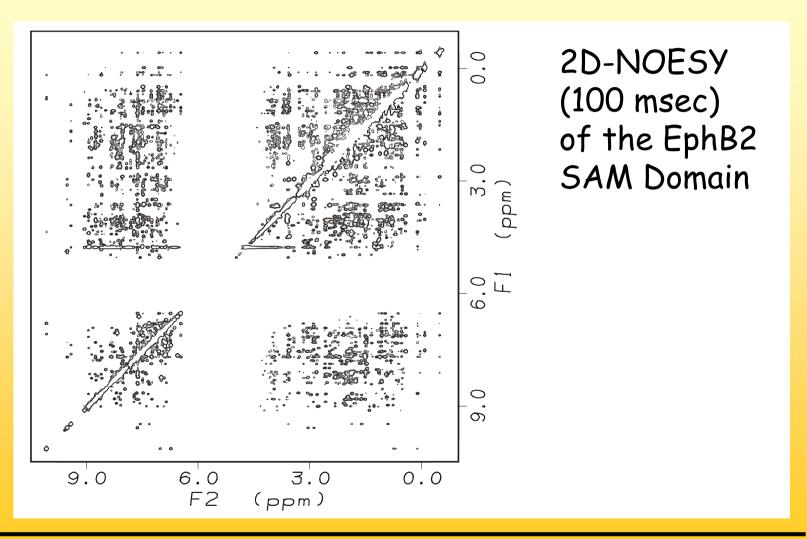
Protein expression and purification









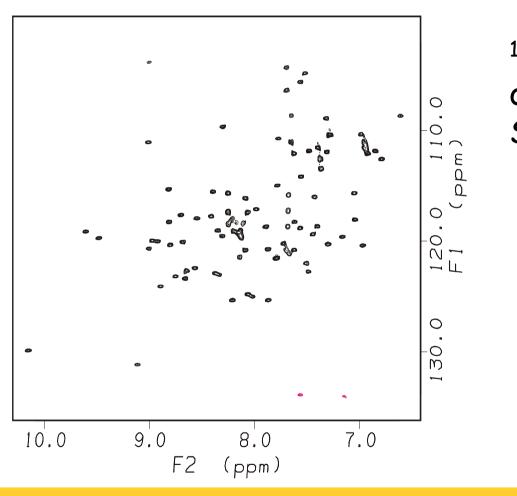




With increasing size of the protein the interpretation of homonucleare spectra alone becomes increasingly difficult.

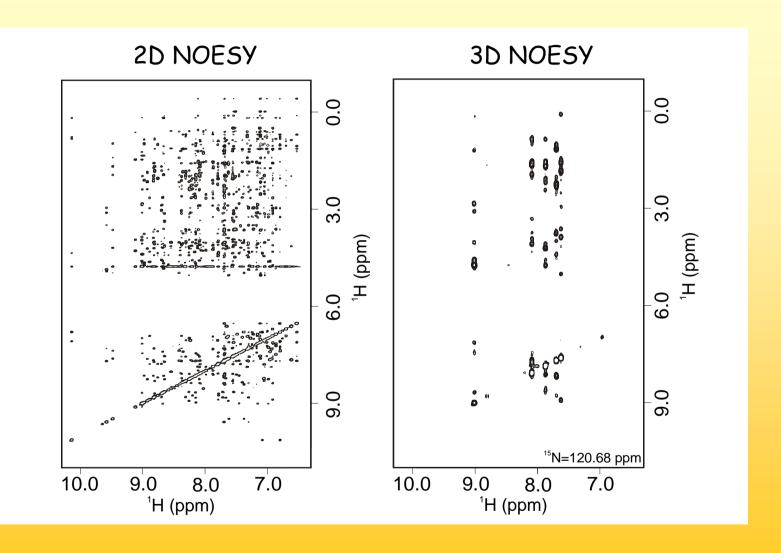
With the introduction of nitrogen and carbon labels this problem can be ameliorated because of the better resolution in the heteronuclear spectra and the option to record well resolved 3D spectra.





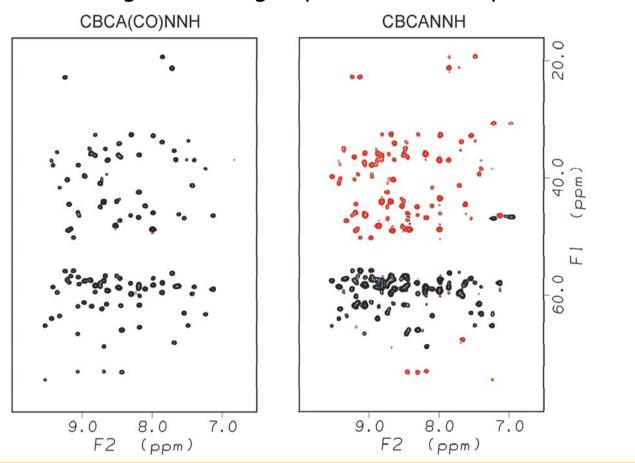
¹⁵N-HSQC of the EphB2 SAM Domain







Mainchain assignment using tripel resonance experiments

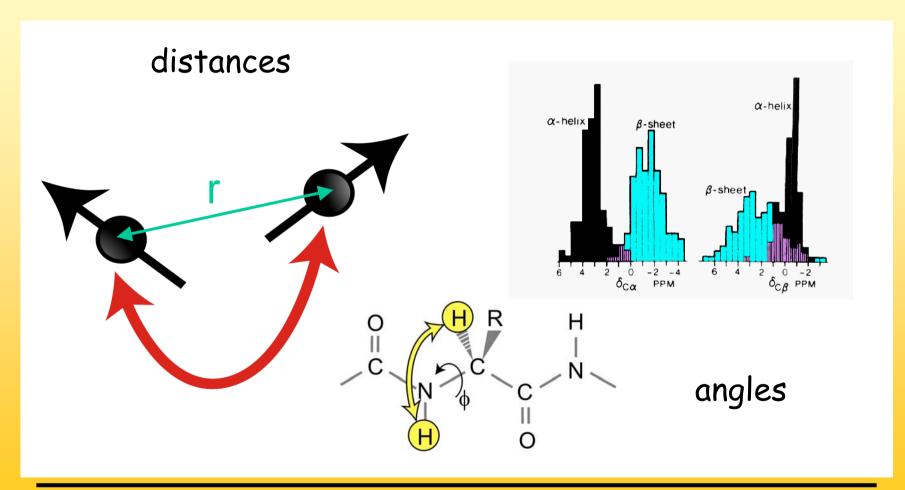




List of relevant experiments ¹⁵N-HSQC, ¹³C-HSQC ¹⁵N-NOESY-HSQC, ¹³C-NOESY-HSQC CBCA(CO)NNH, CBCANNH HNCO, HN(CA)CO HNCA, HN(CO)CA (H)C(CO)NNH, H(CCO)NNH ¹⁵N-relaxation time experiments

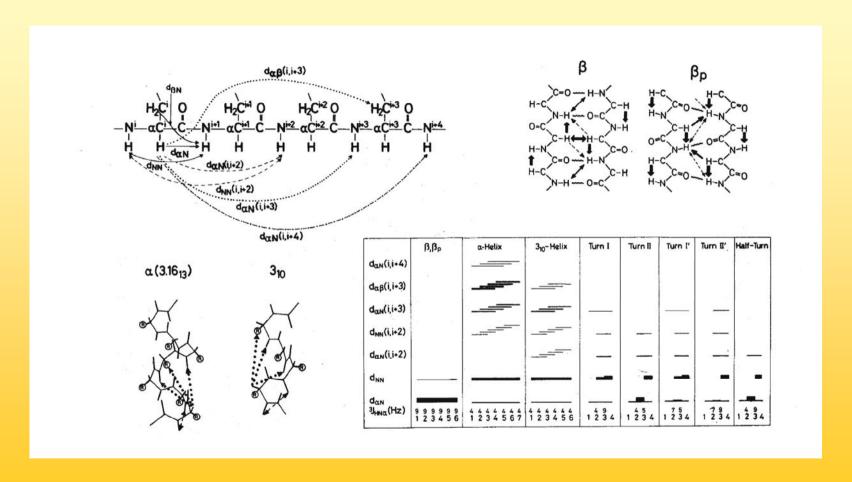


Structurally relevant information





Distances give information on elements of secondary stucture



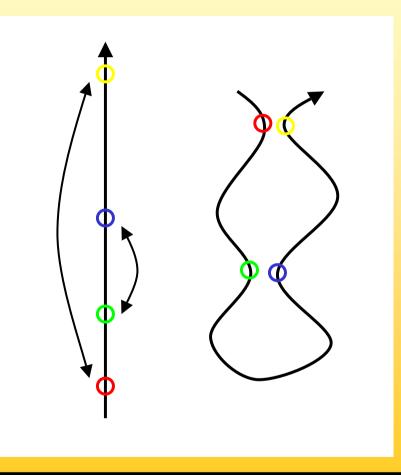


Structurally relevant information





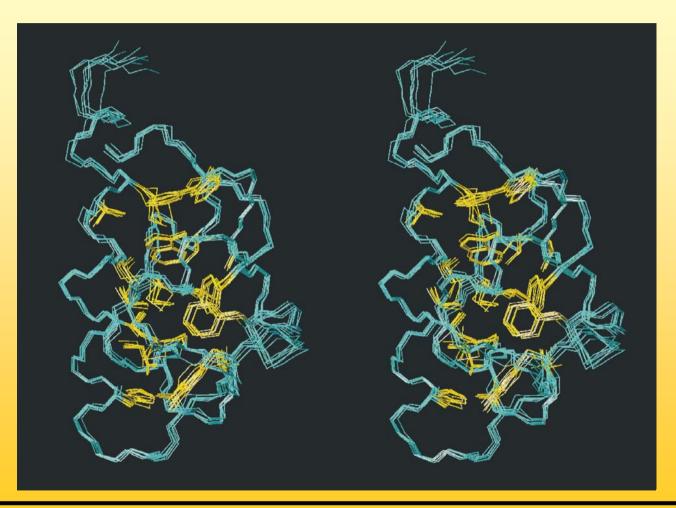
Distances determine the overall structure of the protein



Few distances are enough to "fold up" the protein



As a result a 3D structure can be calculated





Ligand-screening using NMRspectroscopy



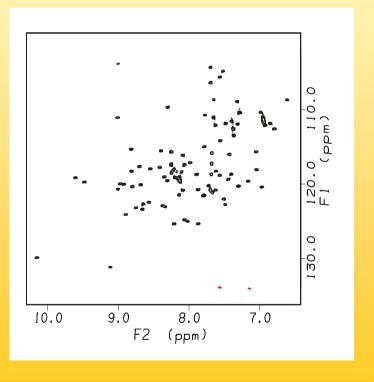
An increasingly important application of NMRspectroscopy is the screening of compound libraries to identify new interaction partners for a given protein and subsequently lead structures

There are two major types of approach, the "ligand-detecting techniques" and the "protein detecting techniques"

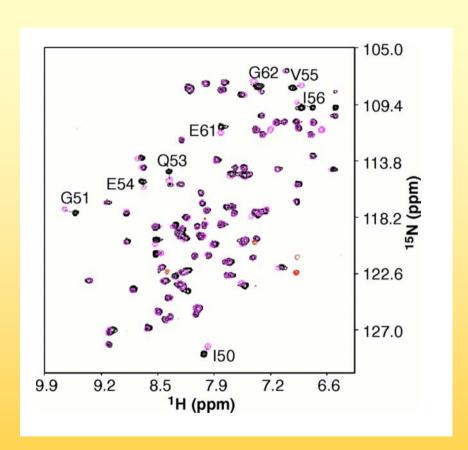


A technique of major importance from the class of protein-detecting techniques is called "SAR-by-NMR"

Starting point is a completely assigned two-dimensional HSQC spectra of the protein of interest





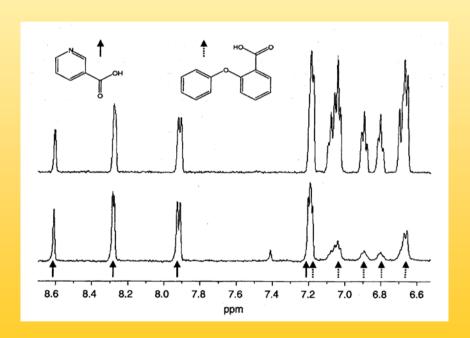


HSQC-specta with and without the addition of a potential ligand are compared. A shift in the spectrum with ligand relative to the one without indicates an interaction

The method can be used in a "high-throughput" manner



There are numerous techniques in the class of the "ligand detecting techniques"
Two techniques of particular importance are WATERlogsy and STD-NMR



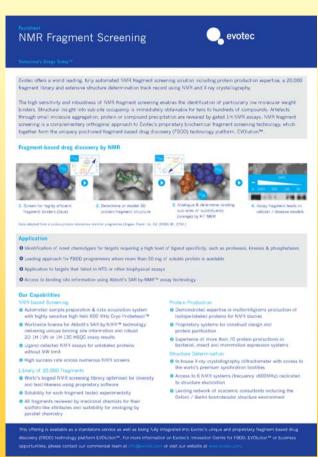
The major principle is
the alteration of certain
properties of the ligand
by the protein when
bound



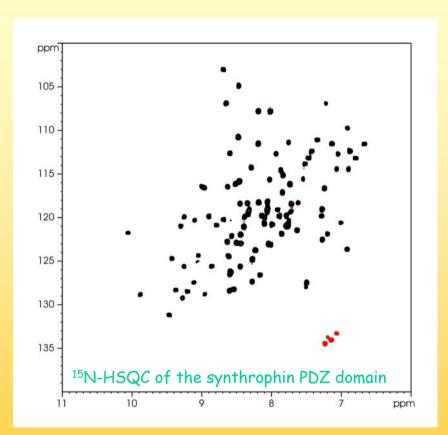
Screening can be commercially interesting....

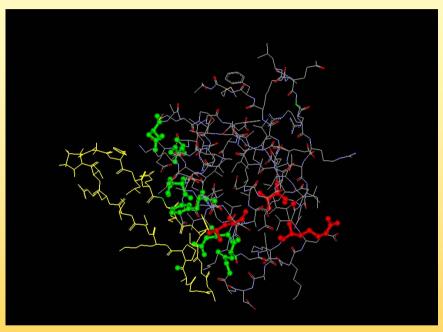












...but it can also be used to detect specific interations



Summary

using protein-NMR-spectroscopy it is possible

to determine the structure of small to medium sized proteins

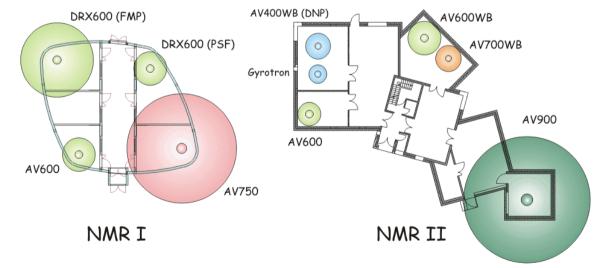
to study protein-protein or protein-ligand interactions



The NMR facility









That's it

www.fmp-berlin.de/schmieder/teaching/educational_seminars.htm

